



**2-Year Master of Science (M.Sc.) Curriculum and  
Syllabus for Chemistry**

**Second Semester**

Course Code	Course Title	Contact Hrs. / Week			Credit
		L	T	P	
<b>Theory</b>					
TIU-PEN-T100	Career Advancement Skill Development	2	1	0	3
TIU-PCH-T102	Physical Chemistry	3	1	0	3
TIU-PCH-T104	Organic Chemistry	3	1	0	3
TIU-PCH-T106	Inorganic Chemistry	3	1	0	3
TIU-PCH-T108	Analytical Chemistry	3	1	0	3
<b>Practical</b>					
TIU-PCH-L104	Organic Chemistry Lab	0	0	3	2
TIU-PPH-L114	Computer Application in Chemistry	1	0	3	3
<b>Sessional</b>					
TIU-PES-S198	Entrepreneurship Skill Development	0	0	2	2
	<b>Total</b>	15	5	6	<b>22</b>



**Semester-II**

**Physical Chemistry**  
**TIU-PCH-T102**

**L-T-P: 3-1-0**

**Credit: 3**

**Module 1**

**Electrochemistry**

Quantitative treatment of Debye-Hückel theory of ion-ion interaction and activity coefficient, applicability and limitations of Debye-Hückel limiting law, its modification for finite-sized ions, effect of ion-solvent interaction on activity coefficient. Debye-Hückel-Onsagar (D-H-O) theory of conductance of electrolyte solution, its applicability and limitations, Pair-wise association of ions (Bjerrum and Fuoss treatment), Modification of D-H-O theory to account for ion-pair formation, Determination of association constant ( $K_A$ ) from conductance data.

**Module 2**

**Surface Chemistry**

**A. Reactions on surfaces:** Adsorption, adsorption isotherms, unimolecular surface reaction, bimolecular surface reactions-reaction between a gas molecule and an adsorbed molecule, reaction between two adsorbed molecules, inhibition and activation energy of such reactions, volcano curve.

**B. Transition state theory of surface reactions:** rates of chemisorptions and desorption, unimolecular and bimolecular surface reaction, comparison of homogeneous and heterogeneous reaction rates.

**C. Micelles:** Surface active agents and their classifications, micellization, factors affecting cmc of surfactants, Thermodynamics of micellization: phase separation and mass action models, micro-emulsions, reverse micelles.

**Module 3**

**Quantum Mechanics**

**A. Fundamentals of quantum mechanics:** Black-body radiation, photoelectric effect, Davison and Germer experiment, Franck-Hertz experiment, Young's double slit experiment; identification of classical and quantum systems, Bohr's correspondence principle with examples, the uncertainty principle.

**B. Operators in quantum mechanics:** Eigenvalues and eigenfunctions, Hermitian operator and its application. Postulates of quantum mechanics, Angular momentum of a one-particle system, and its commutative relations, Ladder operator, Pauli spin operator, Pauli spin matrices-spin



eigenfunctions and their properties, Schrodinger wave equation and its formulation as an eigenvalue problem.

**C. Quantum mechanical treatment on various systems:** Translational motion of a particle, particle in one and three dimensional boxes, harmonic-oscillator, rotational motion of a particle: particle on a ring, particle on a sphere, rigid rotator, step-potential and tunneling, hydrogen atom.

**D. Approximation methods:** Stationary perturbation theory for non-degenerate and degenerate systems with examples, Variation method.

### Books Recommended

1. J. O'M. Bockris, A. K. N. Reddy, Modern Electrochemistry, Vol. 2 A & B, 2<sup>nd</sup> Edition, Plenum Press, New York (1998).
2. Samuel Glasstone, An Introduction To Electrochemistry, Affiliated East-West Press Pvt. Ltd.- New Delhi (2000)
3. A. J. Bard, L. R. Faulkner, Electrochemical Methods: Fundamentals and Applications; 2<sup>nd</sup> Edition (2001), John Wiley & Sons, New York.
4. Y. Moroi, Micelles: Theoretical and Applied Aspects, Plenum Press, New York (1992).
5. P. W. Atkins, Physical Chemistry, 7<sup>th</sup> & 8<sup>th</sup> Editions, Oxford University Press, New York
6. I. N. Levine, Quantum Chemistry, 5<sup>th</sup> Edition (2000), Pearson Educ., Inc. New Delhi.
7. D. A. McQuarrie, J. D. Simon, Physical Chemistry, A Molecular Approach, (1998), Viva Books, New Delhi.



Organic Chemistry  
TIU-PCH-T104

L-T-P: 3-1-0

Credit: 3

**Module 1**

**Pericyclic reactions:** Molecular orbital symmetry, frontier orbitals of ethylene, 1,3-butadiene, 1,3,5-hexatriene and allyl systems. Classification of pericyclic reactions. Woodward-Hoffmann correlation diagrams. FMO and PMO approach, concept of aromaticity of pericyclic transition states. Selection rules and stereochemical aspects of electrocyclic reactions, cycloaddition and sigmatropic shifts. Electrocyclic reactions: conrotatory and disrotatory motions,  $4n$ ,  $4n+2$  and allyl systems. Cycloaddition reactions: antarafacial and suprafacial additions,  $4n$  and  $4n+2$  systems; 2,2 addition of ketenes, 1,3 dipolar cycloadditions and cheletropic reactions. Sigmatropic rearrangements: suprafacial and antarafacial shifts of H, sigmatropic shifts involving carbon moieties, 3,3- and 5,5-sigmatropic rearrangements. Sommelet-Hauser, Cope, Claisen, and aza-Cope rearrangements. Fluxional tautomerism. Ene reaction.

**Module 2**

**Esterification and hydrolysis of ester:** Evidence for tetrahedral intermediate in  $BAC_2$  and  $AAC_2$  mechanisms, steric and electronic effects, the  $AAC_1$  and other pathways involving alkyl to oxygen bond cleavage.

**Module 3**

**A. Electrophilic and nucleophilic aromatic substitution:** Electrophilic aromatic substitution: The Arenium ion mechanism, orientation and reactivity in monosubstituted benzene rings, ortho/para ratio, Ipso substitution. Nucleophilic aromatic substitution: The Aromatic  $S_N^1$ ,  $S_N^2$  and benzyne mechanisms. Reactivity-effect of substrate structure, leaving group, and attacking nucleophiles.

**B. Elimination reaction mechanism:** The  $E_1$ ,  $E_2$ , and  $E_1cB$  mechanisms, Orientation of double bond, Hoffman elimination, Saytzeff elimination, Hoffman versus Saytzeff elimination, Pyrolytic-*syn*-elimination, competition between substitution and elimination reactions.

**Module 4**

**Formation and reactions of enol and enolate:** Enol and enolate, Stable enol, consequence of enolization, Reactions with enols and enolates as intermediate, Stable enolate ions, Preparation of enol ether, Reactions of enol ethers.

**Books recommended**

1. Clayden, Greeves, Warren, Wothers, Organic Chemistry, Oxford University Press, 2001.
2. M. B. Smith, Jerry March, Advanced Organic Chemistry, 5<sup>th</sup> Edition (2001), John Wiley & Sons, New York.
3. Peter Sykes, A Guide Book to Mechanism in Organic Chemistry, 6<sup>th</sup> Edition (1997), Orient



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Longman Ltd., New Delhi.

4. G. S. Zweifel, M. H. Nantz, Modern Organic Synthesis, (2007), Freeman and Company, New York.
5. S. M. Mukherjee, S.P. Singh, Reaction Mechanism in Organic Chemistry, 1<sup>st</sup> Edition (1990), Macmillan India Ltd., New Delhi.
6. T. H. Lowry, K. S. Richardson, Mechanism and Theory in Organic Chemistry, 3<sup>rd</sup> Edition (1998), Addison – Wesley Longman Inc. (IS Edition).
7. S. M. Mukherjee, S. P. Singh, Pericyclic Reactions, MacMillan India, New Delhi.
8. I. Fleming, Pericyclic Reactions, Oxford University Press, Oxford (1999).



**Inorganic Chemistry**  
**TIU-PCH-T106**

**L-T-P: 3-1-0**

**Credit: 3**

**Module 1**

**A. Kinetics and Mechanism of Substitution Reactions:** Nature of substitution reactions; prediction of reactivity of octahedral, tetrahedral and square-planar complexes in terms of crystal field activation energy and structure preference energy; rates of reactions; acid hydrolysis, base hydrolysis and anation reactions.

**B. Electron Transfer Reactions:** Mechanism and rate laws; various types of electron transfer reactions, Marcus-Hush theory, correlation between thermal and optical electron transfer reactions; identification of intervalence transfer bands in solution.

**Module 2**

**Metal Carbonyls and related compounds:** Preparation, structure, and properties: bonding in metal carbonyls, variants of CO bridging, vibrational spectra of metal carbonyls, principal reaction types of metal carbonyls.

**Module 3**

**Chemistry of Lanthanides and Actinides:** Nuclear stability, terrestrial abundance and distribution, relativistic effect, electronic configuration, oxidation states, aqueous-, redox- and complex- chemistry, electronic spectra and magnetic properties, lanthanide and actinide contractions and their consequences, separation of lanthanides and actinides, organo-lanthanoids and actinoids.

**Books Recommended**

1. F. Basalo, R. G. Pearson, Mechanism of Inorganic Reactions, 2<sup>nd</sup> Edn. (1967), Wiley Eastern Ltd., New Delhi.
2. D. F. Shriver, P. W. Atkins, Inorganic Chemistry, 3<sup>rd</sup> Edn. (1999), ELBS, London.
3. F. A. Cotton, G. Wilkinson, Advanced Inorganic Chemistry, 6<sup>th</sup> Edn. (1999), John Wiley & Sons, New York.
4. D. N. Sathyanarayana, Electronic Absorption Spectroscopy and Related Techniques, Universities Press (India) Ltd., Hyderabad (2001).
5. Keith F. Purcell, John C. Kotz, Inorganic Chemistry, W. B. Saunders Com. (1987), Hong Kong.
6. Martin L. Tobe, John Burgess, Inorganic Reaction Mechanisms, Longmans 1<sup>st</sup> Edn. (1999).



**Analytical Chemistry**  
**TIU-PCH-T108**

**L-T-P: 3-1-0**

**Credit: 3**

**Module 1**

**Spectroscopic Techniques**

Theory, Instrumentation and applications of Atomic absorption Spectroscopy, Atomic fluorescence spectrometry, Atomic emission spectrometry, UV-Visible molecular absorption Spectrometry (principles, instrumentation, and application), Molecular luminescence spectroscopy (fluorescence, phosphorescence, chemiluminescence), Concept of Inductively coupled plasma-atomic absorption spectrophotometer, ICPA-AAS (Instrumentation and application).

**Module 2**

**Chemical Sensors and Separation Techniques**

**A.** Principles, types of chemical sensors based on the modes of transductions, Types of chemical sensor based on the chemically sensitive materials (solid electrolyte, gas, semiconductor), Humidity sensors, Biosensors, Electrochemical sensors (Potentiometric sensors, Ion-selective electrodes, Membrane electrodes, Amperometric sensors, Clark and Enzyme electrodes).

**B.** Principles of chromatography, Classification of chromatography, Paper chromatography, Techniques of Column chromatography, Thin layer chromatography, Gas Chromatography, High-performance liquid chromatography, Ion chromatography.

**Module 3**

**Voltammetry and Thermal Analysis**

Linear sweep voltammetry, Anode sweep voltammetry, Cyclic voltammetry, Polarography, Current-Voltage relationship, Theory of polarographic waves (DC and sampled DC (tast) polarograms), Instrumentation, ilkovic equation (derivation excluded), Differential pulse polarography, Qualitative and Quantitative applications. Thermal Analysis: Theory, methodology and application of Thermo Gravimetric Analysis (TGA), Differential Thermal Analysis (DTA), and Differential Scanning Calorimetry (DSC). Principles, techniques, and application of thermometric titration methods, Amperometric titrations.

**Recommended books**

1. D.A. Skoog, Principles of Instrumental Analysis, 5<sup>th</sup> Edition (1998), Saunders College Publishing, Philadelphia, London.
2. D. A. Skoog, D. M. West, F. J. Holler, S. R. Crouch, Analytical Chemistry-An Introduction, 7<sup>th</sup> Edition, (2000), Saunders College Publishing, Philadelphia, London.
3. Nirmalendu Nath, Kakoli Upadhyay, Avinash Upadhyay, Biophysical Chemistry Principles and Techniques, Himalaya Publishing house, New Delhi.
4. J. H. Kennedy, Analytical Chemistry: Principles, 2<sup>nd</sup> Edition (1990), Saunders Holt, London.



5. G. W. Ewing, Instrumental Methods of Chemical Analysis, 5<sup>th</sup> Edition (1978), McGraw Hill Books Co, New York.
6. Modern method of Chemical Analysis, 2<sup>nd</sup> Edition (1976), John Wiley, New York.
  
7. G.D. Christian,, R. L. Pecsok, L. D. Shields, T. Cairns, L.C. Mc William, Analytical Chemistry, 5<sup>th</sup> Edition (1994), John Wiley & Sons, New York.

**Computer Application in Chemistry**  
**TIU-PPH-L114**

**L-T-P: 1-0-3**

**Credit: 3**

**Module 1**

**Programming in C**

Decision making, looping, and control structures. Functions; recursion; arrays; introduction to pointers; character strings; structures and unions, managing input/output operations, formatted I/O, standard library/user-defined functions.

**Module 2**

**Numerical Methods**

Approximations and round off errors, Truncation errors, Determination of roots of polynomials and transcendental equations by Newton-Raphson, Secant and Bisection's method. Backward, Forward, Newton-Divided Difference Polynomial, integration.

**Module 3**

**Applications of MATLAB in Chemistry**

Solving ordinary and partial differential equations using MATLAB. Solver methods for stiff and non-stiff ordinary differential equations. Finite element analysis. Applications in statistical mechanics and chemical thermodynamics, Bayesian statistics, stochastic simulation and parameter estimation. Monte-Carlo simulation. Statistical inference. Time series analysis and ANOVA.

**Recommended Books**

**I. Programming in C**

1. Let Us C, Yashwant Kanetkar
2. Programming in ANSI - C, Balagurusamy, E, Tata McGraw-Hill

**II. Numerical Methods**





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3. Numerical Analysis, S. Ali Mollah

4. Introductory Numerical Analysis, Dutta & Jana

### **III. Applications of MATLAB in Chemistry**

5. K. J Beers, Numerical Methods for Chemical Engineering: Applications in MATLAB, Cambridge University Press, 2007.

### **Organic Chemistry Lab** **TIU-PCH-L104**

**L-T-P: 0-0-3**

**Credit: 2**

**Experiment 1:** Characterization of organic compounds or groups by spectroscopic methods

**Experiment 2:** Separation of aromatic compounds utilizing their physical properties

**Experiment 3:** Isolation of caffeine from tea leaves

**Experiment 4:** Separation, and identification of organic compounds in binary mixtures

**Experiment 5:** Separation and identification of organic mixtures containing up to three components.

**Experiment 6:** Preparation of organic compounds involving several stages, characterization of intermediates and final products by IR and NMR spectroscopy.

**Experiment 7:** Techniques of organic chemistry: Special practical's involving steam distillation, photo-isomerisation and thin layer chromatography etc.

**Experiment 8:** Quantitative analysis of (i) sulfur and (ii) nitrogen.